

James P. Kiernan, P.E. Project Manager Chevron Environmental Management Company 6001 Bollinger Canyon Road Room B1266 San Ramon, CA 94583 Tel (925) 842-3220 jkiernan@chevron.com

April 22, 2024

Peter Campbell Alaska Department of Environmental Conservation Division of Spill Prevention and Response 43335 Kalifornsky Beach Road, Suite 11 Soldotna, AK 99669

Subject: Historical Records Review and Evaluation Former Chevron Kenai Refinery Nikiski, Alaska ADEC File No. 2323.38.040

Dear Mr. Campbell:

As discussed and agreed upon in prior communications with Alaska Department of Environmental Conservation (ADEC), Chevron Environmental Management Company (CEMC) has conducted an extensive historical records search for the Former Chevron Kenai Refinery site in Nikiski, Alaska, including retrieval and review of hundreds of files. Based on that review, CEMC concludes that there was no storage or use of PFAS-containing firefighting foams at the site. Based on this conclusion and given that there were no detections of PFAS in shallow groundwater other than estimated (J-flagged) results below the analytical quantitation limit of 2 parts per trillion (ppt) (as reported in our November 4, 2021 results summary report), there is no justification for further PFAS investigation at the site.

## HISTORICAL RECORDS RESEARCH

Historical site maps ranging from original 1962 Plot Plans to a 1993 Simplified Plot Plan<sup>1</sup> document the location of fire water storage tankage, fire water pumps, and water hydrants; however, there is no evidence PFAS-containing foam was ever present. Thus, the maps and diagrams all support that only water and non-PFAS extinguishing agents were used. Further, no records indicate that any fires (based on inquiry to the local fire department and former employee interviews, including one who was a volunteer firefighter with Nikiski Fire Department from 1974-2000) have occurred at the site<sup>2</sup>. The only records of any foam present at the site (for training or emergencies) are for non-PFAS (fluorine-free) protein-based foam and a dry chemical extinguishing agent, as detailed below. These conclusions are further substantiated by former employee interviews conducted independently but in parallel to the records review. Excerpts from the relevant historical documentation are included as **Attachment 1**.

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# AOF 3% COLD FOAM (AOF 3CF) TECHNICAL DETAILS FOLLOWING ARCHIVES REVIEW

We reviewed comprehensive records of chemicals stored on-site at the time of facility decommissioning after operations ceased in 1991. These records included February 1992 inventories<sup>2</sup> of Aer-O-Foam 3% Cold Foam (noted as in the laboratory and in 5-gallon pails with 19 containers on-site in the boiler building) stored on-site at that time. Additionally, contemporary (circa 1986) product materials for Aer-O-Foam 3% Cold Foam (AOF 3CF) from the manufacturer (National Foam) were identified within company archives explaining AOF 3CF content<sup>3</sup>. Protein foams contain naturally occurring proteins, often derived from animal wastes such as ox blood, as the foaming agents. Thus, the only record of foam stored at the site is of a non-PFAS, cold-weather protein foam (i.e., AOF 3CF). The following additional external documents were reviewed, and all indicated that AOF 3CF is a fluorine-free, protein-based foam that does not contain PFAS:

- *A Firefighter's Guide to Foam* (January 2002) by National Foam<sup>4</sup> on page 19: Depicts the AOF 3CF approved uses in contrast to other firefighting foams and categorizes AOF 3CF as a protein foam.
- An exhibit to recent PFAS litigation<sup>5</sup> included an October 2008 addendum, *Perfluorocarbon-Containing Firefighting Foams and their Use in Firefighting Training in Minnesota* that also confirms (on Table 1, page 21) AOF 3CF (i.e., AOF 3% Cold) is a Class B protein foam.
- ITRC (2022<sup>6</sup>) notes in Figure 1 (reproduced below) of its *Aqueous Film-Forming Foam* fact sheet that protein foams are fluorine-free and do not contain PFAS.

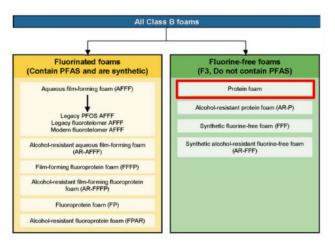


Figure 1. Types of Class B foams. Source: S. Thomas, Wood, PLC. Used with permission. PFAS-1, Figure 3-2.

Based on a review of all available information, AOF 3% Cold Foam, the firefighting foam material identified in the historical site inventories, does not contain PFAS.

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## FIRE TRAINING AREA (FTA) RESEARCH FINDINGS

Known water-only training with a propane torch in June 1981 is included in the archival record<sup>7</sup>. For the period 1970-1980, a 1992 document reported annual diesel and "Special K" [sic]<sup>8</sup> usage at the fire training area (FTA). **Attachment 2** includes additional information from archives regarding FTA operations and extinguishing agents<sup>9</sup>. Though not a "foam", this additional (dry chemical) fire extinguishing agent, Purple K, was also referenced as a firefighting foam in additional references<sup>10</sup> and in former employee interviews, including one with the site manager from the mid-1960s until the mid-1970s, who recalls only dry-chemical fire extinguishers used on-site. Many manufacturers have produced potassium bicarbonate-based Purple K over the years since its original development by the U.S. Navy in 1959, as an improvement over sodium bicarbonate for extinguishing oil and gasoline fires<sup>11</sup>.

Based on one of the former employee interviews, sometime after 1978 (when the employee's tenure started), Chevron had a small, two-wheeled Master Stream fire wagon that could be hooked to the back of a truck or pushed by hand. Former employees do not recall any foam usage beyond AOF 3CF. All available information indicates that no storage, use, or training with any PFAS-containing foam was ever conducted on-site.

## CONCLUSIONS

Based on the available information, the following conclusions regarding the potential for PFAS impacts from any historical operations can be made:

**No PFAS Storage Occurred at the Former Chevron Refinery**: Available information indicates that only non-PFAS foam was stored at the refinery during operation (i.e., before 1991). Fluorine-free AOF 3CF was listed as being on-site in 1992 during the chemical inventories and in 1993 when preliminary waste profiling began during decommissioning. Fluorine-free AOF 3CF is not a PFAS-containing foam.

**No PFAS Usage Occurred During Firefighting at the Former Chevron Refinery**: No records of any PFAS release or use at the site were identified. No record of fire loss or response to any fires was identified based on fire department inquiry, available records, and interviews of long-time employees and local volunteer firefighter experience dating back to 1974.

**No PFAS Was Used in Fire Training at the Former Chevron Refinery**: Usage of dry chemical firefighting agent Purple K and non-PFAS protein foam (AOF 3% Cold Foam) identified at the site is not sufficient evidence to warrant PFAS sampling in site soils, particularly given the absence of any record indicating PFAS-containing foams were ever present or used on-site.

The conclusion from the extensive historical records search that there was **no identified on-site storage or use of PFAS-containing firefighting foams** is also consistent with the August 2021 monitoring well sampling results of *de minimis* estimated (J-flagged) trace concentrations of PFAS (all below the 2 ppt analytical quantitation limit)<sup>12</sup>. Based on this conclusion, there is no basis for requiring CEMC to prepare a PFAS soil investigation workplan, as there is no PFAS "source area" where there are no records of ever having stored or used PFAS-containing materials for firefighting or training purposes. April 22, 2024 Page 4

We appreciate your continued cooperation on this project. If you have any questions on this letter or need any additional information, please contact me at (925) 842-3220 or jkiernan@chevron.com.

Sincerely,

James P. Kiernan, P.E. Project Manager

Cc: Dr. Shanna Clark, CEMC

Attachments:

- 1. Historical Records Excerpts
- 2. Fire Training Area Information

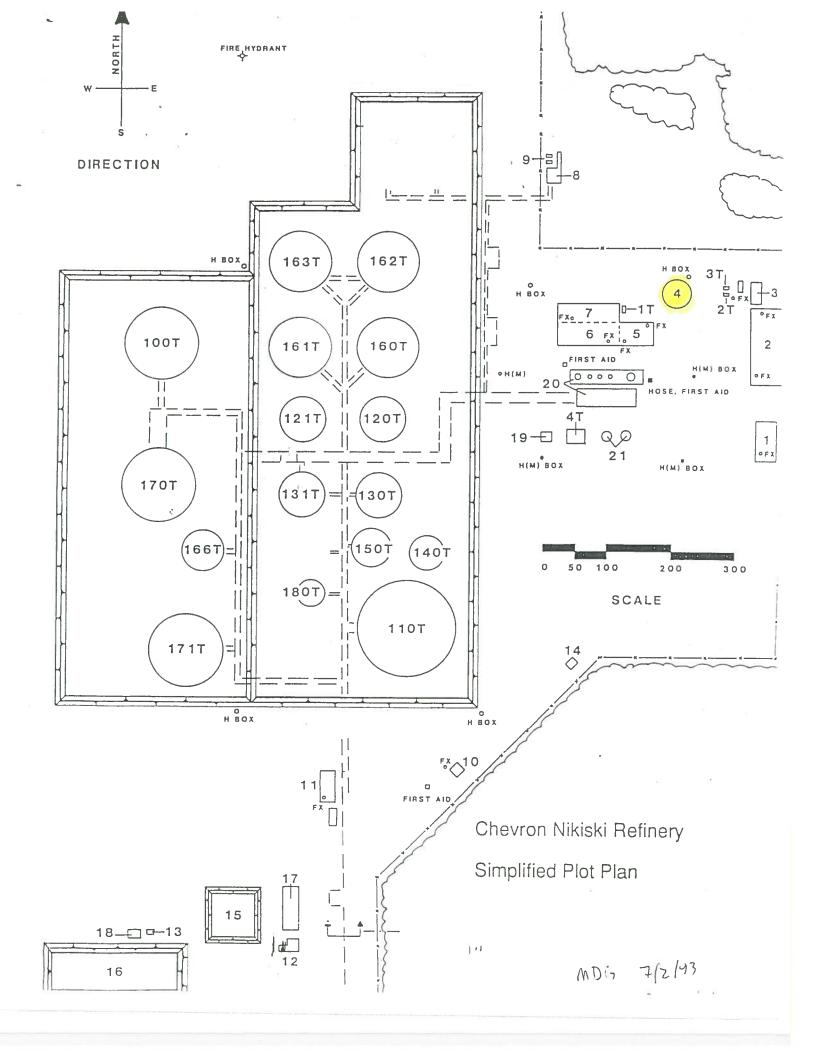
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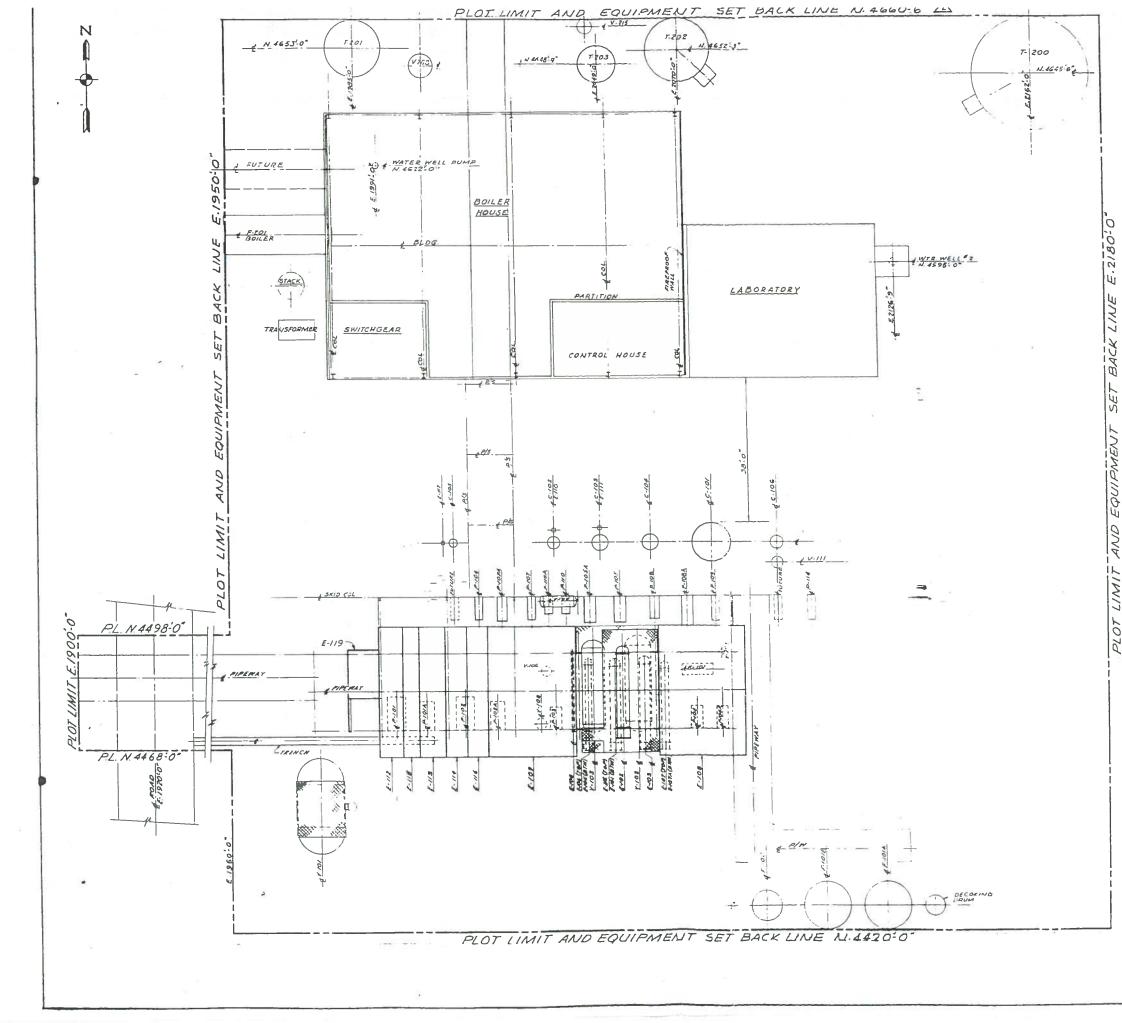
- 1. See Attachment 1, Plot Plan (1962) and Simplified Plot Plan (1992)
- See Attachment 1, Chevron (1992a) 1991 Hazardous Substance Inventory for Refinery: Chemicals Previously Used at the Refinery (dated 2/1/1992) and Chevron (1992b) Hazardous Substance Inventory data as of 2/12/1992
- 3. See Attachment 1, National Foam (circa 1986), Chapter 2 excerpt
- 4. <u>http://www.foamtechnology.us/Firefighters.pdf</u>
- 5. https://www.pca.state.mn.us/sites/default/files/pfc-foamreport-addendum.pdf
- 6. <u>https://pfas-1.itrcweb.org/wp-</u> <u>content/uploads/2022/09/AFFF\_PFAS\_FactSheet\_082522\_508.pdf</u>
- See Attachment 2, Section 4.28 "SWMU 28: Fire Training Area Number 1" (text excerpt from page 49 of USEPA [1992] RCRA Facility Assessment Final Report prepared by PRC, November 23)
- 8. Believed to refer to Purple K, a contemporary dry chemical fire extinguishing product reportedly used (according to former employees) at the site and found elsewhere in the historical archival records.
- 9. See Attachment 2, Section 4.29 "SWMU 29: Fire Training Area Number 2" (text excerpt from page 50 of USEPA [1992] RCRA Facility Assessment Final Report prepared by PRC, November 23)
- See Attachment 2, Section 3.2.23 "Fire Training Area" (text excerpt from page 3-21 of ENSR [1992] Phase II Site Assessment Workplan, August; "RCRA Information Needs" table revised 8/16/92)
- 11. See page 98 of Corbett (2009) Fire Engineering's Handbook for Firefighter I and II.
- 12. See Table 2 of Arcadis (2021). Per- and Polyfluoroalkyl Substances Sampling Results Summary Letter – Former Chevron Kenai Refinery, Nikiski, Alaska. Submitted and received by ADEC on 11/4/21.

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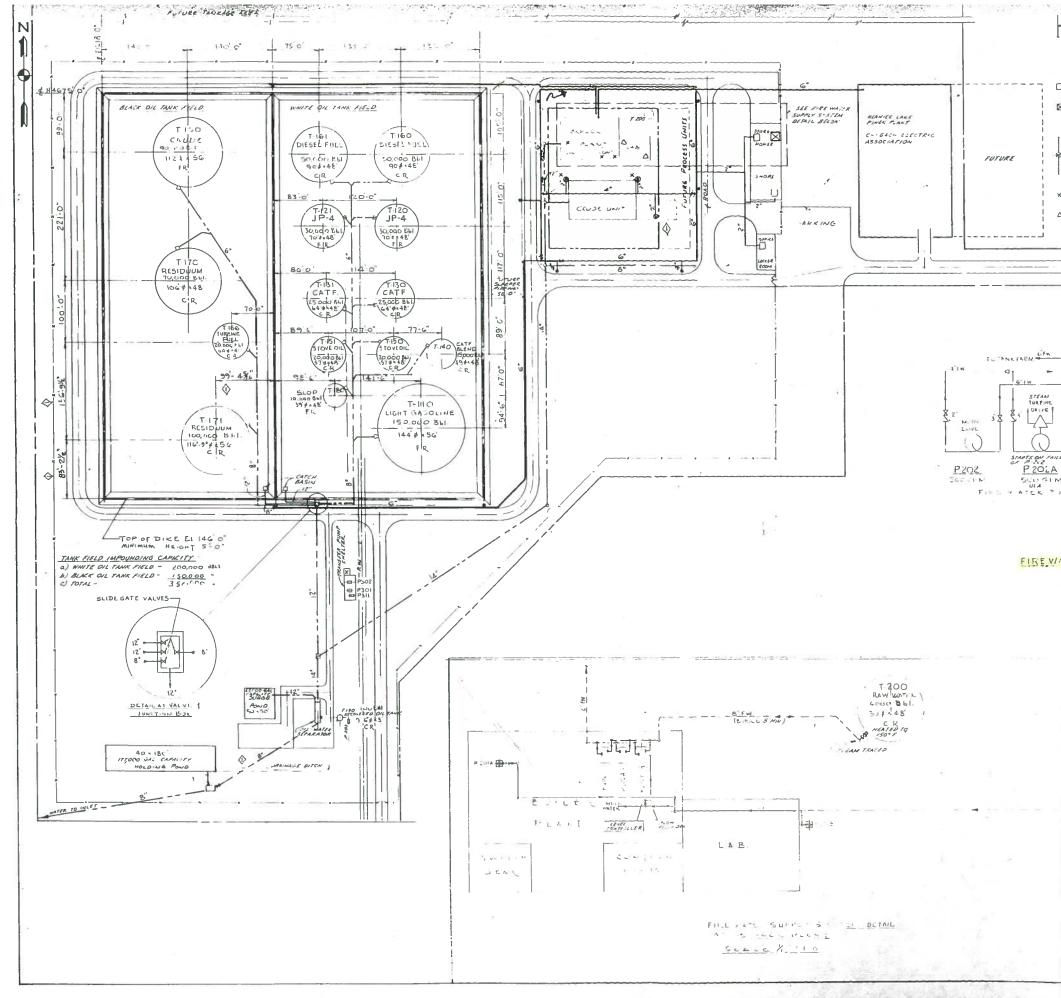
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SUBSTANCE

CHEVRON U.S.A. INC.

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CHEVRON U.S.A. INC. FACHLITY: MAKE - KENAL REFINERY CHEMICAL INVENTORY SYSTEM REPORT 106 - 02/01/92 Inventory by Mork Location

NORK LOCATION NAME	SUBSTANCE NAME	MEGR	CENTRAL HSDS #	SUBSTANCE In
LABORATORY	ERIOCHROME BLACK T	BAKER	X19368	S19360
	ETHYLENE 6LYCOL			CPS310850
	FERROIN			C14708997
	FERROUS AMONIUM SULFATE			C10045893
	FLUDRESCANT REAGENT INDICATOR	COP		\$9492
	FREON 13	VARIOUS		S29804
	FREON 22		YODOLO	C75456
	FULLERS EARTH			C8031183
	GALLIC ACTD		9610BA	C149917
	GLYCEROL (GLYCERIN)		40199	C56815
	HEAVY DILS/RESIDUM	21	002637	S15622
	HELTUN		Y00029	C7440597
	HEPTANE-N		080523	C142825
	HEXANE		001534	C110543
	HYDROBAC MUTANT BACERL HYDRCRB	POLYBAC		S22382
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INC. FACILITY: MAKE - KENAI REFINERY CHEMICAL INVENTORY SYSTEM REPORT 106 - 02/01/92 INVENTORY BY HORK LOCATION CHEVRON U.S.A. INC.

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CHEVRON U.S.A. INC. FACILITY: YOKE - KENAI REFINERY CHEMICAL INVENTORY SYSTEN REPORT 106 - 02/01/92 INVENTORY BY WORK LOCATION

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	ASBESTOS		000302	C1332214
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	CARBON NONOXIDE		000427	C630080
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	CHEVRON ATF SPECIAL	CHEVRON	00000	CPS226587
	CHEVRON DELO 430 N/D SAE 104	CHEVRON	610000	CPS225001
	CHEVRON DELO 400 M/O SAE 15440	CHEVRON	0012100	CPS225006
	DETERGENT	CHEVRON	000869	CPS213502
	CHEVRON DIESEL FUEL NO. 2	CHEVRON	001525	CPS272102
	CHEVRON GEAR OIL SAE 90	CHEVRON	00325	CPS250402
	CHEVRON GST DIL 32	CHEVRON	000221	CPS234229
		CHEVRON	00137	CPS253005
		CHEVRON	000261	CPS250701
	CHEVRON STARTING FLUID SPRAY	CHEVRON	000386	CPS213105
	FX-75 BONDING AGENT	FOX IND		\$35080
	LIDUID AIR COMPRESSED OXYGEN	LIQUID AIR		<b>SLB192</b>
	ROCKWELL LUBRICANT	BTR/ROCK		S2633
	S-C SUPER CONCENTRATE DEGREASE	RADIATOR S		S1526
	WAGNER PREME PLUS BRAKE FLUID	WAGNER CM		S34999
	WEI DINGSCUTTING : FUNES&GASES		092710	S15628
	YELLOW 77 MIRE LUBE	ZDESL		S33718

TOTALS FOR: MAINTENANCE SHP Total Substances 24 Total Substances Mith Chevron MSDS'S: Total Substances Mithdut Chevron NSDS'S:

9 51

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CHEVRON U.S.A. INC. FACILITY: PARE - KENAI REFINERY CHEVRON U.S.A. INVENTORY SYSTEN REPORT 106 - 02/01/92 INVENTORY BY WORK LOCATION

HORK LOCATECH MARE	SUBSTANCE NAME	MFGR	CENTRAL MSDS #	SUBSTANCE TD
OPERATIONS GENERAL		BETZ LABS CHEVRON CHEVRON CHEVRON	000502 000151 000427 000427 000869 000869	59044 C1332214 S18219 C71432 S11126 C630080 CPS213502 CPS213502 CPS213502 CPS213502
	CHEVRON JET LAL 2-C CHEVRON JET FUEL 3-50 CHEVRON NEGILAR GASOLINE CHEVRON UNLEADED GASOLINE CONPRESSED GASES CRUDE OIL DOM ION EXCHANGE RESIN HCRS-NA DUPONT NETAL DEACTIVATOR DHD-2	CHEVRON CHEVRON CHEVRON CHEVRON DOW CH	000545 000545 000372 002703 002703 0022493 K00820	CrS216100 CrS201305 CrS201310 S15626 CrS29600 S18220 S18220 CrS267312
	HAVY DITS/RESIDUM HYDROCARBON LIQUID-EXTREM FLAN HYDROCARBON LIQUID-EXTREM FLAN HYDROCARBON LIQUID-FLANMABLE HYDROEN SULFIDE (H2S) MALCO CUPROUS MALCO CUPROUS MATERS/OILS MASTERS/OILS MASTERS/OILS MASTERS/OILS MASTERS/OILS	MALCD NALCD NALCD CHEVRON MORT THIO CHEVRON CHEVRON CHEVRON	002641 002643 002643 002643 002643 002649 002659 002659 002659 002659	S15622 S15620 S15618 S15618 S15619 C7783064 S15619 S15529 S15629 S15629 S15628 S15628 C75266388 C75266388 C75266388 C75266388 C75266388 C75266388 C75266288 S15628 S15628 S15628

26 TOTALS FOR: OPERATIONS GENERAL TOTAL SUBSTANCES 34 TOTAL SUBSTANCES WITH CHEVRON MSDS'S: TOTAL SUBSTANCES WITHOUT CHEVRON MSDS'S:

60

2

A-8215-03

CHEMICAL INVENTORY SYSTEM REPORT 106 - 02/12/92 FACILITY: MNKE-KENAI REFINERY INVENTORY BY WORK LOCATION

SUBSTANCE 8 CENTRAL **#SDS#** 1-SMALL CYLUDER 1-55gal DRUM 2-35lbs CONTAINER 19-5gal BUCKETS 1/4-55gal DRUM 1-55gal DRUM 6-55gal DRUM 6-55gal DRUMS 432-80lbs BAGS 2-55gal DRUMS 1-55gal DRUM 3-DRUMS VINAUTY **FANK (2340 Ibs)** 1-55gal DRUM 1-55gal DRUM 1-55gal DRUM 1-55gal DRUM 34-80lbs BAGS **FANK (20801bs)** 2-DRUMS 1-DRUM **1 TANK** Sgal 16pints TANK 20002 zsibs Sgal 3lbs gal 000 Sgal Sgal gal d ٩ 2 POLYBAC CORP. MFGR CHEVRON CHEVRON DUPONT SOIL SAMPLES FROM DRILLING OPERATIONS MANGANOUS SULFATE MONOIIYDRATE **GASOLINE (EMERGENCY GENERATOR)** POTASSIUM PHOSPHATE, DIBASIC AER-O-FOAM (3% COLD FOAM) BETZ BALANCED POLYMER 7110 DUPONT METAL DEACTIVATOR DELO 400 MOTOR OIL (SAB 30) ETHYLENE GLYCOL (USED) MISC. PAINTS & SOLVENTS 1,1,1 TCA CHLOROETHENE WASTE OIL (Bast Forelands) CHEVRON GST 32 OIL CHEVRON H.D. CLEANER **BETZ PETROMEEN OS-I6** METHYLENE CHLORIDE HYDROGEN PEROXIDE BETZ MAGNIFORM 304P TRICHLOROBTHYLENE **BETZ MAGNIFORM 304P** SALT (COARSE, SOLAR) HYDROCHLORIC ACID RESIDUAL SAMPLE OIL ACITVATED ALUMINA SAMPLE CONTAINERS SODIUM ITYDROXIDE CHEVRON GST OIL 32 MOLECULAR SIEVE SODYUM BISULFITE SUBSTANCE NAME SALVAGE DRUMS HYDROBAC - S **EMPTY DRUMS UNKNOWN OIL USED DRUMS** SILICONE OIL ACETYLENB BUFFER PH 7 GALLIC AGD 1-BUTANOL SILICA GEL **CITRI PLUS** FREON 13 ACETONE SALT WORK LOCATION NAME MAINTENANCE SHOP NOTE: Not for SARA NOTE: Not for SARA NOTE: Not for SARA BOILER BUILDING reporting. reporting. reporting. LABORATORY TANK 200

PAGE 1

HST Date 2-12-92

A-8215-03

CHEMICAL INVENTORY SYSTEM REPORT 106 - 02/12/92 FACILITY: KENAI PIPELINE FACILITY INVENTORY BY WORK LOCATION

SUBSTANCE A CENTRAL MSDS # 1–351bs CONTAINER 1.5–55gal DRUM 1–5gal BUCKET 2-35lbs CONTAINER 1 TANK (EMPTY7) 1-55gal DRUM 1 TANK **VANTY** 1-55gal DRUM 1-55gal DRUM **JTANK** MFGR CHEVRON CHEVRON CHEVRON H.B. FULLER CHEVRON MOTOR OIL CHEVRON SRI GREASE-2 CHEVRON DETERGENT FOSTER FIBEROUS ADHESIVE MOTOR OIL (DELO 400 PLUS) OXYGEN SHELLZONE ANTIFREEZE SUBSTANCE NAME HYDRAULIC OIL NALCO 5375 LP GAS WORK LOCATION NAME **CRUDE UNIT BUILDING** MAINTENANCE SHOP

**1 TANK** CUPROS ALGAECIDE (COPPER SULFATE) BETZ DX-972 (#4446) IAF BUILDING

**BIODISK BUILDING** 

41/4-Solbs BAGS

PAGE 2



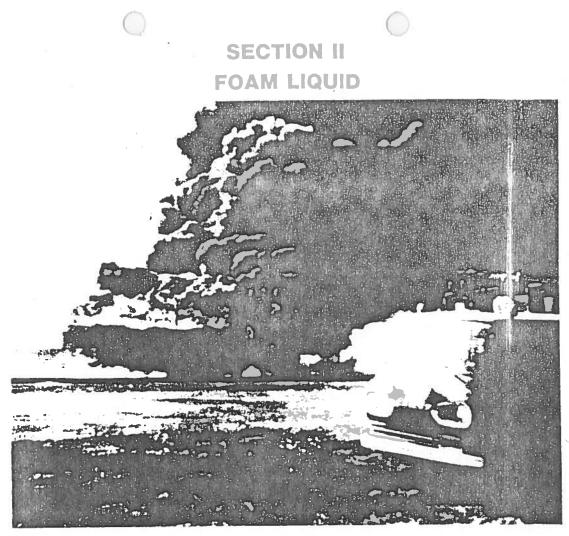


Figure 2-1. National Foam liquid and equipment extinguished this 114 It. diameter crude oil fire in less than 1 hour, saving 50,000 barrels of product, the tank shell, and preventing a "boil over".

#### 2,1 GENERAL

All foam systems, regardless of size, consist of a water supply, a proportioning device, an air aspirating foam maker(s) and a foam liquid supply. While all the components must function properly to assure system performance, the foam liquid is unquestionably the most vital component of the system.

National Foam System, Inc. has been a pioneer in the development of mechanical foam liquids and a world leader in fire fighting foam technology for more than fifty years. National Foam products have extinguished hundreds of flammable and combustible liquid fires, real fires "in anger" involving hazards such as storage tanks, process areas, marine tankers, loading facilities, and spills resulting from aircraft and automotive crashes. A performance record unparalleled in the industry. In every corner of the world, there are National Foam liquids and equipment, some in service for more than thirty years — month after month, year after year — providing reliable, proven flammable liquid fire protection.

#### 2.2 CHOOSING A FOAM LIQUID

With more than one thousand different flammable liquid materials being manufactured and consumed; and with more than twelve separate fire fighting foams available for their protection — the choice of

the most suitable foam to protect a particular hazard may appear difficult. However, this choice becomes objective and measurably simplified when dealing with a manufacturer that produces every type of foam liquid and all the appliances required for their use. National manufactures the longest and most comprehensive line of fire fighting foams available to the industry. Whether the hazard is a small marketing installation or a sophisticated storage and process facility. National's diverse product line simplifies selection of the best, most cost-effective foam liquid.

Basically, there are two general classes of foam liquid; the regular protein based type and the synthetic type. Within each class are modified forms which provide a specific foam to best meet the requirements of a particular hazard.

#### **Protein Based Types**

Aer-O-Foam 3% Regular Aer-O-Foam 6% Regular Aer-O-Foam 3% Cold Foam Aer-O-Foam 6% Cold Foam Aer-O-Foam XL-3 3% Fluoroprotein Aer-O-Foam XL-6 6% Fluoroprotein Aer-O-Foam XL-6 6% Fluoroprotein Cold Foam Aer-O-Foam "99" 6% Alcohol Resistant

#### Synthetic Types

Aer-O-Water 6 - 6% AFFF Aer-O-Water Plus - 3% AFFF (Available in Cold Foam) Aer-O-Water PSL --- 6%-10% AFFF and Alcohol Resistant Universal - 3%-10% Multi-Purpose High Expansion --- 11/2 %-3% Syndet

Basically, there are two general classes of flammable liquids; hydrocarbons and polar solvents. Hydrocarbons are non-water miscible products such as crude oil, gasoline, hexane, naphtha, diesel oil, etc. Polar solvents are generally water miscible products such as alcohols, esters, ketones, etc. Some industrial solvents are a mixture of both classes.

The following information should be available for

- c indideration in order to properly choose the most s. .table foam liquid:
- 1. Principal flammable liquids requiring protection (actual chemical title).
- 2. Foam solution application rates (determines water supply requirements).
- 3. Foam liquid cost.
- 4. System components and field piping cost.
- 5. Projected cost of foam system maintenance.

A few dollars in foam liquid cost can save tens of thousands of dollars in field piping and maintenance expense. Conversely, protection systems can be over-designed around costly foam liquid, when a

less expensive foam will provide totally acceptable protection. Table 2-1 lists foam liquids, the hazards they protect, and methods of application.

#### 2.3 TESTING AND APPROVALS

All National Foam products undergo extensive testing from their conception in the research laboratory through rigid quality control standards prior to market. Foam liquid that is physically and chemically stable assures a long storage life and optimum fire performance. In this regard, National's foam liquid products are approved and listed by independent testing agencies such as Underwriters' Laboratories and Factory Mutuals. Certain liquids are also approved by the U.S. Coast Guard and other Federal agencies. These approvals are the customers' guarantee that National has demonstrated through extensive fire testing and evaluations that the product complies with the rigid requirements and specifications of the testing authority. We note further, any deviation from these standards can lead to a revocation of said listings or approval. In special cases, fire tests are conducted to determine the effectiveness of the foam on a particular flammable liquid and to compute the minimum application rates the hazard requires. NFPA Standards 11,4 11B, 16 and 409 provide the guidelines for determining application rates. Some application rates for polar solvent or alcohol type fuels are determined by the foam liquid manufacturer through actual fire testing. The approvals for each particular foam liquid are provided in their descriptive paragraphs.

## Table 2-1. Recommended Foams, Proportioning % and Application Methods for Various Hazards

FOAM LIQUID	HYDROCARBON HAZARDS (Reference — N.F.P.A. Standards 11 & 11B)			POLAR SOLVENT (ALCOHOL) HAZARDS		
	Storage Tanks (5 .10 GPM/Fl <sup>2</sup> (4 LPM/r	See Section VI) m <sup>2</sup> ) Application Rate	Spill Fires — .16 GPM/Ft² (6 LPM/m²)	Storage Tanks (See Section VI)	Spill Fires Nozzles, Monitors, Overhead Devices	
	N.F.P.A. Type II Fixed Topside Chambers	Subsurface Injection	Nozzies, Monitors, Overhead Devices	on Applicat	-2 for Details ion Rates and r Requirements	
Fluoroprotein			e -			
Aer-O-Foam XL-3	3%	4%	3%	NR	NR	
Aer-O-Foam XL-6 (& Cold Foam)	6%	6%	6%	NR	NR	
Regular Protein	541 1					
Aer-O-Foam 3% (& Cold Foam)	3%	NR	3%	NR	NR	
Aer-O-Foam 6% (& Cold Foam)	6%	NR	6%	NR	NR	
Aer-O-Water (AFFF)	-				<u> </u>	
Aer-O-Water Plus (& Cold Foam)	3%	NR	3% *	NR	NR	
Aer-O-Water 6	NR	NR	6%*	NR	NR	
Universal	3%	4%	3%*	6%-10% (Type II)	6%-10%	
Aer-O-Water PSL	6%	NR	6%*	6%-10% (Type II)	6%-10%	
Aer-O-Foam "99"	6%	NR	6%	6% (Type I)	6% (Skin Spills Only)	

\*N.F.P.A. 11B allows a minimum application rate of .10 gpm/ft<sup>2</sup> (4 lpm/m<sup>2</sup>) for nozzle applications.

NR = Not Recommended



Figure 2-2. Proven again in action. National's XL-3 readily extinguished this gas-oil intermediate fire involving two 45 ft, diameter tanks and their common dike area. Foam-dry chemical truck built by National in 1968.

#### 2.4 NATIONAL AER-O-FOAM XL FLUOROPROTEIN LIQUIDS

XL Fluoroprotein foam liquids represent the single most significant improvement in foam technology since mechanical foams were first introduced. By the combination of selected fluorocarbon surfactants with a quality protein hydrolysate base, a foam liquid with dramatically improved performance characteristics is produced. These improved characteristics include:

- 1. Increased extinguishment ability
- 2. Increased fluidity
- 3. Dry chemical compatibility
- 4. Superior sealability and burnback resistance

5. Vastly improved olephobic properties These olephobic properties are so pronounced that they permit subsurface injection of XL foams into hydrocarbon storage tanks. (See Section VI. Storage Tank Protection.)

Since its development in 1965, Aer-O-Foam XL has repeatedly established itself as the best foam agent available for subsurface and topside storage tank protection. In its history there have been no performance failures in real emergencies and no reported failures due to natural deterioration in storage. The performance record of XL foams is in fact so unparalleled that more of the world's major oil and chemical companies rely on National Aer-O-Foam XL for protecting their flammable hydrocarbon products, than on any other foam.

XL Fluoroprotein foams are available for 3% and
6% proportioning and are suitable for use with fresh or sea water.

4.1 General Properties
er-O-Foam XL-3
pecific Gravity @ 60°F (15.5°C) 1.158 to 1.164
H 7.1 to 7.5
iscosity in Centistokes
@ 20 <sup>2</sup> F (-6.6 <sup>2</sup> C)
linimum Usable Temperature 20°F (-6.6°C)
aximum Usable Temperature 120°F (48.8°C)
ecommended Maximum Storage
Temperatures 100°F (37.7°C)
lecommended Storage Container
Material Mild Steel
pprovals Underwriters' Laboratories
U. S. Coast Guard
Factory Mutual
New York Board of Standards
SBG — Germany

#### Aer-O-Foam XL-6

Specific Gravity @ 60°F (15.5°C) 1.147 to 1.153
pH 7.1 to 7.5
Viscosity in Centistokes
@ 20°F (-6.6°C) 200 csks
Minimum Usable Temperature 20°F (-6.6°C)
Maximum Usable Temperature 120°F (48.8°C)
Recommended Maximum Storage
Temperatures 100°F (37.7°C)
Recommended Storage Container
Material Mild Steel
Approvals U. S. Government
Underwriters' Laboratories

#### Aer-O-Foam XL-6 Cold Foam

Specific Gravity @ 60°F (15.5°C) 1:11 to 1.15
pH 7.1 to 7.5
Vis Dosity in Centistokes
(1 – 20°F ( – 29°C) 1200 csks
Minimum Usable Temperature 20°F (- 29°C)
Maximum Usable Temperature 120°F (48.8°C)
Recommended Maximum Storage
Temperatures 100°F (37.7°C)
Recommended Storage Container
Materials Mild Steel
Approvals Underwriters' Laboratories

## 2.4.2 Approximate Shipping Weights

5 Gallon (19 litre) Pails

NOTE: Please indicate packaging preference on purchase order.

## 2.5 NATIONAL'S AER-O-FOAM REGULAR LIQUIDS

Three types of Aer-O-Foam Regular Liquids are available. These liquids are manufactured from pure protein hydrolysate, compounds for foam stabilization, freezing point depressants. and preservatives. All are carefully blended to produce a homogenous. highly stable foam liquid. Regular Aer-O-Foams are designed for use on hydrocarbon type flammable liquid fires through N.F.P.A. Type II devices and air aspirating foam nozzles. Available in both 3% and 6% concentrations. Regular Liquids can be used with fresh or sea water. Special "cold foams" are available for use in frigid climates or where heating of the foam liquid in storage is not feasible.

#### 2.5.1 General Properties

Aer-O-Foam 3% Regular
Specific Gravity @ 60°F (15.5°C) = 1 158 to 1.164
pH
Viscosity in Centistokes
@ 20°F (-6.6°C) 400 csks
Minimum Usable Temperature 20 F (-6.6°C)
Maximum Usable Temperature 120 F (48.8°C)
Recommended Maximum Storage
Temperature 100°F (37.7°C)
Recommended Storage Container
Material Mild Steel
Approvals Underwriters' Laboratories
U. S. Coast Guard
Factory Mutual
SBG Germany
Norsk Veritas — Norway
Norwegian Maritime Directorate

Aer-O-Foam 6% Regular	
Specific Gravity @ 60°F (15.5°C	) 1.139 to 1.145
рН	· · · · · · · · 7.1 to 7.5
Viscosity in Centistokes	
@ 32°F (0°C)	80 csks
Minimum Usable Temperature .	20°F (-6.6°C)
Maximum Usable Temperature .	120°F (48.8°C)
Recommended Maximum Storag	
Temperatures	100°F (37,7°C)
Recommended Storage Containe	er
Material	Mild Steel
Approvals	. U. S. Government
	U.S. Coast Guard
	Factory Mutual
Aer-O-Foam 3% Cold Foam	
Specific Gravity @ 60°F (15.5°C	) 1.11 to 1.15
pH	7.1 to 7.5
Viscosity in Centistokes	
@ −20°F (−29°C)	1200 csks
@ - 20°F (- 29°C) Minimum Usable Temperature .	
<ul> <li>@ - 20°F (-29°C)</li> <li>Minimum Usable Temperature .</li> <li>Maximum Usable Temperature .</li> </ul>	
<ul> <li>@ - 20°F (-29°C)</li> <li>Minimum Usable Temperature .</li> <li>Maximum Usable Temperature .</li> <li>Recommended Maximum Storag</li> </ul>	
<ul> <li>@ - 20°F (-29°C)</li> <li>Minimum Usable Temperature .</li> <li>Maximum Usable Temperature .</li> <li>Recommended Maximum Storag Temperatures</li> </ul>	1200 csks 20°F (- 29°C) 120°F (48.8°C) ie 100°F (37.7°C)
<ul> <li>@ - 20°F (-29°C)</li> <li>Minimum Usable Temperature .</li> <li>Maximum Usable Temperature .</li> <li>Recommended Maximum Storag Temperatures</li> <li>Recommended Storage Contain</li> </ul>	1200 csks 20°F (-29°C) 120°F (48.8°C) ie 100°F (37.7°C) er
<ul> <li>@ - 20°F (-29°C)</li> <li>Minimum Usable Temperature .</li> <li>Maximum Usable Temperature .</li> <li>Recommended Maximum Storag Temperatures .</li> <li>Recommended Storage Contain Material .</li> </ul>	1200 csks 20°F (-29°C) 120°F (48.8°C) ie 100°F (37.7°C) er Mild Steei
<ul> <li>@ - 20°F (-29°C)</li> <li>Minimum Usable Temperature .</li> <li>Maximum Usable Temperature .</li> <li>Recommended Maximum Storag Temperatures</li> <li>Recommended Storage Contain</li> </ul>	1200 csks 20°F (- 29°C) 120°F (48.8°C) je 100°F (37.7°C) er Mild Steei writers' Laboratories

3% and 6% Cold Foams to  $-40^{\circ}$ F ( $-40^{\circ}$ C) are available upon request.

## 2.5.2 Approximate Shipping Weights

NOTE: Please indicate packaging preference on purchase order.

#### 2.6 NATIONAL AER-O-WATER LIQUIDS (AFFF)

Aqueous Film Forming Foams (AFFF) were developed by the U.S. Navy in the middle 1960's, AFFF's are a combination of fluorocarbon surfactants and synthetic foaming agents that add a new dimension to crash rescue fire fighting: the aqueous film. This film is a thin layer of foam solution that rapidly spreads across the surface of a hydrocarbon fuel causing dramatic fire "knock down." an important factor in crash rescue fire fighting. The aqueous film is produced by the action of the fluorocarbon surfactant reducing the surface tension of the foam solution to a point where the solution can actually be supported by the surface tension of the hydrocarbon fuel. The effectiveness and durability of the aqueous film is directly influenced by the surface tension of the hydrocarbon. AFFF's are more effective on fuels with higher surface tension coefficients such as kerosene, diese: oil and jet fuels: less effective on fuels with low surface tension coefficients like hexane and high octane gasolines. AFFF foams are constructed to drain foam solution quickly from the foam bubble to produce optimum filming for rapid fire extinguishment. Long term sealability and burnback resistance are sacrificed by this rapid drainage.

National's contributions to the development of AFFF's has been significant. The greatest development to date was the introduction of Aer-O-Water Plus; designed for 3% proportioning, and patented by National Foam. By design, Aer-O-Water Plus requires a very low enegry input to produce a good quality foam and a long lasting blanket. When used through aspirating foam nozzles, the foam shows increased extinguishment ability and burnback resistance. Aero-O-Water Plus demonstrates such pronounced foam making ability that it will produce foam from non-aspirating equipment which has an impinging stream action to help entrain air. When used through certain water fog nozzles, Aer-O-Water Plus produces quality foam with overall improvement in fire "knockdown" and control when compared to conventional AFFF. Of even greater significance, Aer-O-Water Plus produces foam from standard water sprinkler heads. The foam produced is of sufficient quality to meet the requirements of Underwriters Laboratories, Inc. for AFFF, and is listed without limitation for use with most major standard sprinkler heads. Aer-O-Water lus will also produce adequately expanded foam through many directional and spray nozzles used in special hazards systems. Existing water deluge systems are readily converted to Aer-O-Water Plus foam systems by merely adding the appropriate proportioning equipment.

Aer-O-Waters can be used with fresh or sea water, and Aer-O-Water Plus "Cold Foam" is available upon request. All Aer-O-Waters perform as excellent wetting agents on "Class A" fires, and are compatible with dry chemical agents.

2.6.1 General Properties
Aer-O-Water Plus
Specific Gravity @ 77°F (25°C) 1.05 to 1.07
pH
Viscosity in Centistokes
@ 77°F (25°C) 15 csks
Minimum Usable Temperature 25°F (-3.8°C)
Maximum Usable Temperature 120°F (48.8°C)
Recommended Maximum Storage
Temperatures 100°F (37.7°C)
Recommended Storage Container
Material Mild Steel (%) 304 Stainless Steel
Approvals Underwriters' Laboratories Factory Mutual
Conforms to fire performance
of MIL-F-24385
Aero-O-Water Plus is available as a Cold Foam - 20°F ( - 29°C)
Aer-O-Water 6
Specific Gravity @ 60°F (15.5°C) 1.02 to 1.04
pH 7.4 to 8.0
Viscosity in Centistokes

Viscosity in Centistokes
@ 77°F (25°C) 4 csks
Minimum Usable Temperature 30°F (-1.1°C)
Maximum Usable Temperature 120°F (48.8°C)
Recommended Maximum Storage
Temperature 100°F (37.7°C)
Recommended Storage Container
Materials Mild Steel (3/6")
304 Stainless Steel
Approvals Military Specification MIL-F-24385
Underwriters' Laboratories

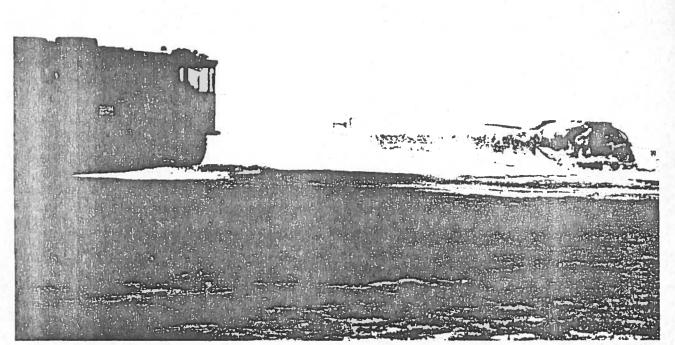


Figure 2-3. 7200 sq. ft. of JP-4 spill fire. Totally extinguished by Aer-O-Water Plus In 37 seconds. Application rate .07 gpm/ft.<sup>2</sup> (3 lpm/m<sup>2</sup>).

Synthetic foam liquids may require additional tests to evaluate surface tension, viscosity and the effectiveness of the aqueous film or polymeric membrane.

The Technical Service Report will list the results of all these tests. If the foam liquid sample produces results consistent with is original specifications, it is considered satisfactory and suitable for fire service. Significant deviation from the original specifications in any of the test results usually indicates one of the following problems:

- 1. Contamination
- 2. Improper Storage Procedures
- 3. Microbial Decomposition
- 4. Product Degradation
- 5. Any combination of the above

 as point, fire testing is recommended. The actual firs performance of a foam sample will determine its suitability for fire service.

#### 2.11.2 Fire Tests

Various size tests can be devised depending upon agent and characteristics being evaluated. A test commonly employed is a modification of a Federal Specification (OF-555C) for Mechanical Foam Liquids intended for storage tank protection and industrial use.

A foam liquid is scored satisfactory if the fire is extinguished within 5 minutes application time. The resulting foam blanket must form a seal against reignition for 15 minutes when tested with a flaming torch. A void is then cut in the blanket and ignited. The opening must not enlarge significantly within an additional 5 minutes burn period.

Alcohol or polar solvent type foams are similarly fire tested on a polar fuel such as isopropanol. Aer-O-Foam "99" is tested via a Type I application, while Universal and Aer-O-Water PSL use a Type II method. AFFF (Aer-O-Water Plus and Aer-O-Water 6) are fire tested according to test procedures in Federal Military Specification MIL-F-24385.

## 2.11.3 Sample Collection

Obviously, the foam sample submitted must be representative of the foam liquid storage, whether in tanks or drums. Proper foam liquid sampling is of concern to the analyst in a twofold way. First, the original sample collected from storage and submitted for analysis, and secondly, the proper sampling of the submitted sample for test purposes. Depending upon the maintenance program adopted, foam liquid samples are collected in one of the following ways:

- 1. Collect one sample-bottom only
- 2. Collect two samples --- one top and one bottom
- Collect three samples one each from top, bottom and middle
- Collect one sample composite after recirculating contents

All samples should be submitted in a clean one pint (500 ml) plastic bottle or steel can. If fire testing is recommended a larger sample will be requested. "Request for Analysis" forms are available from our local representative or from our home office.

#### **Bottom Sampling**

Since the bottom of the tank may collect sediment such as rust, scale, or degradation products, it is important that these excessive contaminants be separated when drawing a sample.

A suggested procedure is as follows:

- Open the bottom-most drain and flush out one or two gallons of liquid into a large clean bucket.
- Close down tightly on the valve and collect at least one pt (500 ml) in a clean plastic bottle. This is the sample to be submitted for analysis.
- 3. The initial flushings may be returned to the top of the tank by way of the filling funnel. A strainer will remove the excess sediment.

#### **Composite Sampling**

Composite sampling is a good technique to use where the number of samples being collected would get out of hand. However, the contents of a storage tank must not be recirculated where dilution is known or suspected to have occurred. Many fire protection engineers employ a periodic start-up of their pumps and recirculation of the foam liquid as part of their required maintenance program. Samples collected from the drain valve will, of course, be treated as bottom samples.

2.11.4 The Importance of Foam Liquid Sampling It should be emphasized here, that if samples of foam liquid are sent to National's Technical Service Department on a regular basis, problems involving storage conditions can usually be detected and corrective measures recommended before the foam liquid is irreversibly damaged. Periodic sampling of foam liquid assures its ready status in a fire emergency.

## 2.11.5 Additional Technical Services

National's Technical Service Department is not limited to foam liquid analysis. While fire fighting foams find their main application in extinguishing flammable liquid fires, there are many other uses for foam liquids. As petrochemical processes and the products they produce become increasingly sophisticated, more and more applications for foam are being explored by the industry. Unique hazards require unique foams and special methods of application. Many non-flammable chemicals release caustic or toxic vapors. Specialized foams can be provided as vapor or fume suppressing agents. Some chemicals with vapor or flammability problems are violently reactive with water, the principal content of finished, expanded foam. Special techniques can be developed to attack these hazards. Our chemists and technicians are among the most knowledgeable and experienced in the industry. A modern laboratory complete with field test facility is devoted solely to the development and testing of foam agents. We are proud of National's record as a pioneer in the flammable liquid fire protection field, and are anxious and prepared to solve new problems born of advancing technology.

4.28

# SWMU 28 - FIRE TRAINING EXERCISE AREA NUMBER 1

## Unit Description

A fire training exercise is described in ADEC documents (1981a, 1981b). During the VSI, Chevron representatives described the nature and location of this fire training exercise as follows (PRC 1992b). A lit propane torch was used in conjunction with safety equipment to increase employee confidence in safety equipment. This training occurred at the southeast corner of the CRU. This is described as a small event, and no flammable materials were applied to soils.

## Dates of Operation

This training exercise occurred once in June 1981.

## Wastes Managed

Propane and water were used during this exercise. Wastes generated were gaseous and are expected to have dissipated by now.

History of Release/Release Controls

No release controls were used, although release to air may have occurred. An air permit was obtained from ADEC for a fire training exercise (ADEC 1981b).

## Information Needs

The exact location of this SWMU should be provided on a map. No releases from this unit are expected to remain, so no further testing is recommended at this unit.

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4.29

#### Unit Description

According to Chevron (1992d), this SWMU was an unlined pit south of the CRU. Every year, it was flooded with 50 gallons of diesel and water and set a fire. The fire was put out as a fire training exercise.

## **Dates of Operation**

This training exercise was held annually from 1970 to 1980.

#### Wastes Managed

Some diesel and fire extinguishing material may remain in the soil matrix.

#### History of Release/Release Controls

No release controls are documented. An air permit was obtained from ADEC for a fire training exercise (ADEC 1981b). Activities at this SWMU probably resulted in a release to soils as well as air. Chevron plans surface soil sampling from this area (ENSR 1992c).

## Information Needs

- Results from planned sampling and analysis of soils at this SWMU
- Description of any backfilling or soil removal which may have occurred

## 4.30 SWMU 30 - DISPOSAL AREA SOUTH OF CRU (Photograph 64)

#### Unit Description

A flat, graveled area used for disposal of facility equipment is south of the CRU and within the facility perimeter fence. This area was not mentioned in the Hart Crowser (1992a) report and may contain materials from the facility dismantling process.

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## 3.2.22 Above-ground Fuel Tanks

## Area Description

This area consists of the former locations of two, 300-500 gallon tanks of regular leaded and unleaded gasoline. The tanks were installed above a concrete containment pad and were surrounded by an 18" thick concrete wall.

## Potential and Existing Sources/Types of Contamination

The potential that soil and groundwater contamination sourced from this area is highly unlikely.

## Planned Course of Action; Phase II Assessment

No further sampling is planned in this area.

## 3.2.23 Fire Training Area

## Area Description

This area was formerly an earthen pit in which fire-training exercises involving the use of diesel fuel were conducted. Once a year the pit was filled with water, and a 55 gallon drum of diesel fuel was pumped onto the surface of the water. This fuel was then ignited, and subsequently extinguished using "Special K" extinguisher (probable sodium bicarbonate). The water would then infilter into the ground, usually within a few hours.

## Potential and Existing Sources/Types of Contamination

Unburned diesel fuel and dissolved hydrocarbons may have infiltered into soil at this location. No samples have been collected to date.

## Planned Course of Action: Phase II Assessment

Samples will be collected from trench SS-5, following location of pit via airphotos analysis.

## 3.2.24 Crude Refining Unit

## Area Description

This area is a complex, consisting of distillation towers, heat exchangers, pumps, a desalter, and furnaces. The crude unit is located on a concrete pad.

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# RCRA Information Needs (Cont'd) SWMUs/AOCs on Chevron Alaskan Refinery Kenai, Alaska

SWMU/AOC/ PAOC	Location	Unit Description (Narrative)	Dates of Operation	Operational Status	Waste Types	Waste Quantities	Waste Sources	Waste Disposition	Release Controls	Remarks
SWMU 26	Fire training area.	Unlined pit which was annually flooded with water and 50 gallons of diesel fuel. Fuel was ignited and fire was then extinguished using "Special K" (probably sodium bicarbonate).	Circa 1970 to 1980.	Abandoned.	Diesel, fire extinguisher.	On the order of 500 gallons of diesel fuel.	Fuel drums.	Still in place.	None	Pit was infilled after abandonment.
AOC 1	Wastewater treatment system (excluding designated SWMUs).	Boiler blowdown, east of 1601.	1963 lo 1991.	inactive.	Unknown.	Unknown.	Unknown.	Unknown.	None.	None.
AOC 2	Crude refining unit.	Series of pumps, heat exchangers, desaiter, and furnaces.	1963 lo 1991.	inactive.	Crude oil and waler.	Unknown.	Water stations, steam knockouts, and pumps.	Routed to API separator.	Monitored by operators.	<ol> <li>Current concrete slab south of unit was not originally installed.</li> <li>Prior to Installation of slab extension, a large spill occurred south of the crude refining unit, east of the desatter.</li> </ol>
AOC 3	Pipeline from dock to refinery	Numerous oil, water, steam, and condensate lines.	1963 lo 1991.	Inactive.	Oil, refined products, and water.	Unknown.	Pipe leaks.	Still in place.	Spills recovered with vacuum trucks.	<ol> <li>Possibly damaged by 1964 earthquake.</li> <li>Two known leak incidents on the KPL site; under first bridge south of Chevron property ' and at stanchion #11.</li> </ol>

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R Revised 8/16/92

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